

# Study the effect of chromium complex on the graphitization of phenolic resins for developing thermally stable nano carbon fillers

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## Abstract.

The research aims to study the effect of  $\text{Cr}(\text{C}_5\text{H}_7\text{O}_2)_3$  as a promoter to accelerate the graphitization of phenolic resins for developing thermally stable nanocarbon fillers to prepare advanced insulators for petrochemical industries. Two types of matrix precursors were used in the present study; the non-modified and modified phenolic resins. Modified resin is prepared through a chemical reaction of resin with an inorganic chromium complex. Graphitization has been successfully achieved in a vacuum at a temperature not exceeding  $1400^\circ\text{C}$ . The effects of  $\text{Cr}(\text{C}_5\text{H}_7\text{O}_2)_3$  on the graphitization degree of nanoparticles carbon were studied by XRD, FTIR, TG, TGA, and DTG. The infra-red results showed that the  $\text{Cr}(\text{C}_5\text{H}_7\text{O}_2)_3$  complex exhibited an intense band at  $3400\text{ cm}^{-1}$  related to (OH) stretching vibration, and the hydrogen of the hydroxyl replaced by metal ions at ( $287.4\text{-}354\text{ cm}^{-1}$ ), related to (M–O). On the other hand, the  $\text{Cr}(\text{C}_5\text{H}_7\text{O}_2)_3$  form adducts with polymers containing functional groups such as phenols. X-ray analysis showed the  $\text{Cr}(\text{C}_5\text{H}_7\text{O}_2)_3$  complex promoted the formation of more perfect crystallites, which related to graphitic and turbostratic structures (T&G- components). Thus, the catalyst's action is to accelerate the graphitization process and reduce its degree to  $1400^\circ\text{C}$  compared to that needed by amorphous carbon in the conventional process of about  $2800^\circ\text{C}$ .